Total Synthesis of (+)-Asperlin

Subban Ramesh and Richard W. Franck*

Department of Chemistry, Hunter College/CUNY,

695 Park Ave., New York, NY 10021 (Received 22 December 1989)

Absrracr: A stereochemicaIly unambiguous synrhesis of (+)-asperlin from L-rhamnose establishes the configuration of the antibiotic as 4S. 5S, 6S, 7R.

Asperlin (1), a crystalline metabolite from Aspergillus nidulans, has been shown to exhibit antitumor and antibacterial activity. The gross srructure of 1 was assigned by Argoudelis in 1966,' and early synthetic studies by Perlin demonstrated that the C-4 and C-5 substituents of the lactone ring had the L-three configuration. Perlin also proved the *trans* stereochemistry of the epoxypropyl sidechain.² However the absolute stereochemistry of the sidechain of this deceptively simple molecule has eluded a definitive assignment. Thus, NMR studies³ were used to propose the absolute stereochemistry of the sidechain cpoxidc as 6R,7S, an assignment that contradicted the X-ray analysis which assigned the 6S,7R configuration. " Rabanal" and, very recently, Shing" have synthesized unnatural diastereomers of asperlin and deduced the absolute stereochemistry by exclusion. In fact, there has been only one synthesis of (+)-asperlin reported in the literature'; and the method used to prepare the epoxide did not establish its absolute configuration. In this communication, we wish to report an enantiospecific synthesis of 1 which furnishes incontrovertible evidence for its absolute stereochemistry as 4s. 5s. 6S, 7R. We chose to use a sugar precursor that had the requisite stereochemistry at the future C-Q and C-6, that combined with unambiguous chemical transformations, leaves no doubt about C-S and C-7. As illustrated in the retrosynthesis scheme 1, we envisioned the epoxide as arising from 2 and an intramolecular Homer-Wadsworth-Emmons reaction producing the lactone. The appropriate precursor for 3 was L-rhamnose, since it is readily available and needs an inversion only at C-3.

L-Rhamnose was converted to methyl rhamnoside using Binkley's procedure: The manuever that left the C-3 OH free and blocked the C-2 and C-4 OH's was performed as follows. The C-2 and C-3 OH's were protected as an orthoacetate and the C-4 OH was silylated with TBDMSOTf to afford 6 (Scheme 2). The crude silyl ether-otthoacetate was hydrolyzed with 90% aqueous CH,COOH, following well-established precedents for the regioselective ring opening of axial-equatorial orthoesters.' We thus obtained 7 in 52% overall yield from methyl rhamnoside. With the C-3 OH unblocked, its inversion was the next problem to be solved. Although Mitsunobu conditions were unsuccessful in our hands, the transformation was accomplished cleanly in two steps: PCC-MS oxidation' of the alcohol 7 gave the ketone 8 which was stereoselectively reduced to the axial alcohol 9 in 94% overall yield.⁸ The alcohol was esterified to give the phosphonate 10 in 95% yield.¹⁰ Conventional methods (e.g. aqueous acid) when employed to hydrolyze the methyl glycoside led to deblocking of TBDMS and/or phosphonate groups. The hydrolysis was efficiently done in two steps:(i) the methyl glycoside was converted to the 1-chloro sugar 11 by reacting it with BCl, in CH,CI, at -78°C and (ii) the I-chloro sugar was reacted with Ag,CO, in aqueous acetone to give 3 in 73% overall yield. The stage was set for the intramolecular HWE reaction. Several bases and reaction conditions were tried¹² but NaH/THF at room temperature gave satisfactory results to produce the lactone I2 in 32% yield. The lactone was converted to (+)-asperlin as follows: the remaining free OH was converted to its mesylate; then reaction of the mesylate with n-Bu,NF.3H,O in CH,Cl, at room temperature" produced asperlin in 22% yield". This material was identical to a sample of the natural product in all respects.¹⁵ Synthetic (+)-Asperlin: mp = $68-70^{\circ}$, $[\alpha]_p25^{\circ}$ =+322° (c= 0.2, 95% EtOH); lit.mp 71-73", [a],25"=+345 (c= 0.9, 95% EtOH)'; 'H NMR (CDCI,, 300MHz) 6 7.11 (dd, lH, J= 5.78, 9.77, H3), 6.26 (d, IH, J= 9.74, IIZ), 5.35 (dd, 1H. J= 5.73, 2.82, HJ). 4.14 (dd, IH,J= 6.95, 2.81, 115), 3.14-3.09 (m, 2H. H6, H7), 2.18 (s, 3H,OAc), 1.43 (d, 3H, J= 5.04. CH,); 13C NMR (CDCl,. 75MHz) 6 169.8 (COCH,); 161.5 (Cl) ; 140.5 (C3); 124.9 (CZ); 78.9 (CJ); 62.1 (CS); 54.9 (C6); 54.6 (C7); 20.6 (CH,CO); 17.0 (C8).Thus, we have achieved a total synthesis of (+)-asperlin which unambigously establishes the absolute stereochemistry of the epoxide as drawn in 1.

Acknowledgement This research was supported by NJH grant CA 37359 and PSC/CUNY research grant program.

Keferences and Notes

I.(a) A.D. Argoudelis, J.H. Coats and R.R. Herr, Antimicrob. Agents and Chemother, 1965, 801; S.P. Owen and B.K. Bhuyan, ibid; 1965, 804. (b) A.D. Argoudelis and J.F. Zieserl, Tetrahedron Lett. 1966, 7. 1969.

2. S. Lesage and A.S. Perlin, Can. J. Chem., 1978, 56, 2889; Idem. ibid; 1978, 56, 3117. These early workers used a numbering system where the lactone carbonyl was designated as C-l.

3.(a) P. Dais and A.S. Perlin, Can. J. Chem., 1985. 63, 1009.(b) K. Fukuyama, Y. Katsube, A. Noda, T. Mamasaki and Y. Hatsuda, Bull. Chem. Soc. Jpn, 1978, 51, 3175.

4.(a) S. Valverde, B. Herradon, R.M. Rabanal and M. Martin-Lomas, Can. J. Chem., 1987, 65, 339.

- (a) MeOH, Dowex-H⁺, heat (b) (EtO)₃CCH₃, DMF, cat. p-TosH, heat
- (c) TBDMSOTf, 2, 6-lutidine, CH_2Cl_2 , $0^{\circ}C(d)$ aq. acetic acid (90 %)
- (e) PCC-MS, $3A^{\circ}$ MS, NaOAc, CH_2Cl_2 , RT (f) NaBH₄ / MeOH, -78 °C
- (g) (EtO)₂POCH₂CO₂H, DCC, CH₂Cl₂, RT (h) BCl₃, CH₂Cl₂, -78[°]C
- (i) Ag₂CO₃, aq. acetone, RT (j) NaH / THF, RT
- (k) MsCl, $E_{13}N$, 0° to RT (l) Bu₄NF. 3H₂O, CH₂Cl₂, RT

4.(b) T.K.M. Shing and M. Aloui, J. Chem.Soc., Chem. Commun. 1988, 1525.

5. T. Murayama, T. Sugiyama and K. Yamashita, Agric. Biol. Chem., 1987, 51, 2055. For a racemic synthesis see: H. Hiraoka, K. Furuta, N. Ikeda and H. Yamamoto, Bull. Chem. Soc. Jpn.. 1984, 57, 2777.

6. R.W. Binkley, G.S. Goewey and J.C. Johnston, J. Org. Chem., 1984, 49, 992.

7.(a) J.F. King and A.D. Allbrett, Can. J. Chem., 1970, 48, 1756 and references cited therein; (b) S. Hanessian and E. Moralioglu, *ibid*, 1972, 50, 233.

8. J. Herscovici and K. Antonakis, J. Chem. Soc.. Chem. Commun., 1980, 561; J. Herscovici, M.J. Egron and K. Antonakis, J. Chem. Soc., Perkin Trans I, 1982, 1967.

9. O. Han and H. W. Liu, Tetrahedron Lett., 1987, 28. 1073.

10. A. Hassner and V. Alexanian, Tetrahedron Lett., 1978, 19, 4475: B. Neises and W. Steglich, Angew. Chem. Int. Ed. Eng., 1978, 17, 522.

11. G. R. Perdomo and J.J. Krepinsky, Tetrahedron Lett., 1987, 28, 5595. We have independently shown that BCI, in CH,Cl, at -78°C converts methyl glycosides to 1-chloro sugars and that -0Ac. -TBDMS and phosphonates are stable under these conditions.

12.(a) S.F. Donovan, M.A. Avery and J.E. McMuny. Tetrahedron Left.. lY7Y, 20, 3287: (b) M.A. Blanchette. W. Choy, J.T., Davis, A.P. Essenfeid, S. Masamune, W.R. Roush and T. Sakai, Tetrahedron Lett., 1984, 25, 2183. Other bases used: KOtBu/ THF; KHMDS/THF; NaHMDS/THF; LDA/THF.

13. J. Kuwajima, T. Murojushi and E. Nakamira, Synthesis. IY72, 602; J. Mayami, N. Uno and A. Kaji, ibid., 1068. 1385.

14.(+)-Asperlin is not particularly stable to Bu,NF. Thus. no other characterizable epoxylactone could be detected in, or isolated from the reaction mixture.

15. All numbered compounds except for 6 and II were purified by chromatography to homogeneity and every compound gave 'H and ''C NMR consistent with the assigned structure. For 10: $[\alpha]_D$ 25°=-51.9° (c= 2.72, CH₂Cl₂), 'H NMR (CDCl₃, 300MHz) δ 4.97 (t, J₂, and J₃₄ = 3 Hz, H3); 4.80 (d,J = 3, H2); 4.42 (bs, H1); 4.1 (m, CH₁OP); 3.96 (m, H5); 3.58 (dd. $J_{4,5} = 9$, $J_{3,4} = 3$, H4); 3.27 (s, OMe); 2.91 (ABq with added coupling to P, CH,-P); 2.01 (s, CH,CO);I.25 (1, CH,): 1.15 (d, CII,); 0.77 (s, t-Bu); 0.00 (s. MeSi); - 0.2 (s, MeSi); "C NMR (CDCI,, 75MHz) δ 169.1 (COCH,); 165.0 (COCH,P); 98.1 (C1); 70.8, 69.9, 69.6 (CZX3,CS); 64.4 (CZ); 62.5,62.6 (CH,OP); 55.1 (OMe); 33.9 (d, J=135, CH,P); 25.5 (Me of t-Bu); 20.8 (Cl&CO); 17.8 (qC of t-Bu); 17.7 (C6);16.3,16.2 (CH,CH,OP); -4.6,4.9 (MeSi). M/e calcd. for $C_{11}H_{41}O_{10}P\text{Si}: 512$; Found (CI-methane) M+1 513. For 12: $[\alpha]_{10}25^{\circ}$ =+112.8° (c= 1.95, CH₂Cl₂), [']H NMR (CDCl,, 300MHz) 6 7.27 (dd, J=9.6,6.0. 113); 6.19 (d, J=9.6, H2); 5.15 (dd, J= 6.0,1.9, HJ); 4.41 (dd, J= 8.8,1.9, 115); 4.20 (m, H7); 4.18 (dd, J= 8.8, 2.2, H6); 2.12 (s, CH₃CO); 1.29 (d, CH₃CH); 0.93 (s, t-Bu); 0.21.0.09 (MeSi); "C NMR (CDCI,, 75MHz) 6 169.9 (COCH,); 162.2 (CO-lactone); 141.5, 124.6 (C=C); 78.1, 72.6, 68.2, 62.4 (C4-C7); 25.7 (Me of t-Bu); 20.8 (CH₃CO); 18.3 (qC of t-Bu); 16.8 (C8);-4.2.-5.1 (MeSi). M/e Calcd. for C16H2806Si: 344; Found (CI-methane) M+l 345.